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Section 15-5 Spontaneous Processes

<ul><li>A spontaneous process is a physical or chemical change that once begun, occurs with no outside intervention. </li></ul><ul><li>Many spontaneous processes require some energy from the surroundings to start the process. </li></ul> 53.

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Chemistry: Matter and Change Chapter 15: Solutions Chapter 19: Acids and Bases. Terms in this set (75) boiling point elevation. The temperature difference between a solution's boiling point and a pure solvent's boiling point. Brownian motion.

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science. the science of different kinds of

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## **Chapter 15 Answer Key ...**

Figure 15.4.1 Energy Changes

Accompanying the Thermite Reaction

Because enthalpy is a state function, the overall enthalpy change for the reaction of 2 mol of Al(s) with 1 mol of Fe<sub>2</sub>O<sub>3</sub>(s) is -851.1 kJ, whether the reaction occurs in a single step ( $\Delta H$  4, shown on the left) or in three hypothetical steps (shown on the right) that involve the successive formation of solid Al<sub>2</sub>O<sub>3</sub> and ...

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