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Spell. Test. PLAY. Match. Gravity. Created by. akapell. Use this quiz to test your skills in writing chemical equations. Terms in this set (127)  $\text{Na} + \text{F}_2 \rightarrow \text{NaF}$ . Sodium reacts with fluorine to form sodium fluoride.  $\text{K} + \text{I}_2 \rightarrow \text{KI}$ . Potassium reacts with iodine to form potassium ...

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A balanced chemical equation allows one to determine A. The mole ratio of any two substances in the reaction B. The energy released in the reaction C. The electron configuration of all all elements in the reaction D. The mechanism involved in the reaction

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## **Chapter 9 Prep-Test**

Chapter 9 Practice Test Multiple Choice Identify the letter of the choice that best completes the statement or answers the question. \_\_\_\_ 1. A solid produced by a chemical reaction in solution that separates from the solution is called a. a precipitate. c. a molecule. b. a reactant.

## **ch\_9 practice test - Chapter 9 Practice Test Multiple ...**

296 Chapter 9 • Chemical Reactions. Figure 9.15 When aqueous sodium hydroxide is added to a solution of copper(II) chloride, the anions ( $OH^-$  and  $Cl^-$ ) change places. The resulting products are sodium chloride, which remains in solution, and copper(II) hydroxide, the blue solid in the test tube.

## **Chapter 9: Chemical Reactions**

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## Answers

Study Guide for Chapter 9: Chemical Reactions 9.1 Reactions and Equations 5 physical changes that indicate that a chemical reaction might have taken place Translating word equations into skeleton (unbalanced) equations Balancing chemical equations Using state symbols (s), (aq)... 9.2 Classifying Chemical Reactions 5 types of chemical reactions

## Honors Chemistry Study Guide for Chapter 9: Chemical Reactions

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Hi students, in this video we will learn to solve numericals of chemical kinetics based on first order reactions. Watch the video carefully. Pl. Like, share and comment.

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